Synthesis, alkylation, and molecular and crystal structure of N-methylmorpholinium 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane-2-thiolate

V. D. Dyachenko, S. G. Krivokolysko, V. N. Nesterov, Yu. T. Struchkov, t and V. P. Litvinovc*

^aT. G. Shevchenko Lugansk State Pedagogical Institute, 2 ul. Oboronnaya, 348011 Lugansk, Ukraine

^bA. N. Nesmeyanov Institute of Organoelement Compounds, Russian Academy of Sciences, 28 ul. Vavilova, 117813 Moscow, Russian Federation.

Fax: 007 (095) 135 5085

^cN. D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, 47 Leninsky prosp., 117913 Moscow, Russian Federation.

Fax: 007 (095) 135 5328

The reaction of cyano(cyclohexylidene)thioacetamide with cyanothioacetamide or malononitrile and N-methylmorpholine yields N-methylmorpholinium 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane-2-thiolate. Its structure was established based on the results of alkylation and X-ray structural analysis.

Key words: cyano(cyclohexylidene)thioacetamide, cyanothioacetamide, malononitrile; X-ray structural analysis.

Recently, the chemistry of functionally substituted hydrogenated pyridines has drawn the attention of researchers because of the biological activity of these compounds. Among them, compounds that exhibit cardiovascular, 1,2 hypertensive, 3-5 and other kinds of activities have been found. Spiro derivatives of hydrogenated pyridines have a special place in this series. This field of the chemistry of heterocycles remains poorly studied, primarily because of the lack of convenient methods for preparing these compounds. However, they represent a promising class of heterocycles from the standpoint of a search for new biologically active compounds. 6

As part of continuing studies devoted to the synthesis and investigation of the properties of hydrogenated 3-cyanopyridine-2(1H)-chalcogenones,⁷ we studied the reaction of cyano(cyclohexylidene)thioacetamide (1) with cyanothioacetamide (2) in the presence of an excess of N-methylmorpholine. N-Methylmorpholinium 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane-2-thiolate (6) was obtained in a yield of 71% (method A). Compound 6 was also obtained in a yield of 78% by the reaction of compound 1 with malononitrile (3) in the presence of N-methylmorpholine (method B). Both reactions proceed apparently as the Michael addition to yield adducts (4) and (5), respectively. As a result of subsequent intramolecular cyclocondensation (with

elimination of hydrogen sulfide, method A) under the action of a base (N-methylmorpholine), adducts 4 and 5 transform into salt 6. The structure of 6 has been established by X-ray structural analysis (Fig. 1; Tables 1 and 2). When treated with 10% HCl, N-methylmorpholinium salt 6 gave 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclohexane-2-thiol (7) in a yield of 92%. The melting point and spectral characteristics of 7 differ from those of thione, which has been obtained previously⁸ by the reaction of cyclohexylidenemalononitrile with 2 or by the reaction of cyano(cyclohexylidene)thioacetamide with 3 and to which the structure of 6 was ascribed. X-ray structural study of salt 6 cast some doubt on the data reported previously.⁸

According to the data of the 1H NMR spectra, compound 7 occurs in a DMSO solution as a mixture of prototropic tautomers (7A) and (7B) in a ratio of 1:1. Thus, the signal at δ 4.60 (s, C(3)—H) confirms the structure of thione (7B), whereas the signal at δ 4.06 (br.s, SH) confirms the structure of its tautomer (7A). It is interesting that the prototropic tautomerism is also typical of the derivatives of thione 7, namely, of 2-alkylthiopyridines (9) and thienopyridines (11) (Scheme 1). In this case, amino and imino forms, 9A, 11A, and 9B, 11B, are observed in the 1H NMR and IR spectra. Thus, the IR spectra of pyridines 9 showed absorption bands of stretching vibrations of the conjugated cyano group in the region of 2180-2200 cm⁻¹ and absorption bands of stretching vibrations of the non-

[†]Deceased.

Bond

C(2)-C(3)

C(2)-C(7)

C(3)-C(4)

C(3)-C(7)

C(7)-N(2)

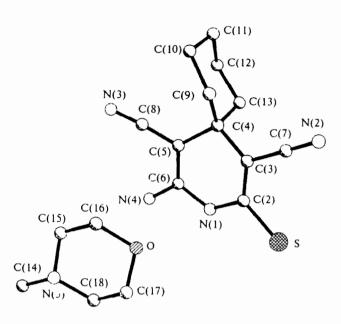


Fig. 1. Overall view of molecule 6.

Table 1. Bond lengths (d) and bond angles (ω) in the structure of 6

Bond

C(8)-N(3)

C(5)-C(6)

C(6)-N(4)

C(6)-N(1)

N(1)-C(2)

d/Å

1.154(4)

1.371(3)

1.348(3)

1.378(3)

1.385(2)

d/A

1.363(3)

1.737(2)

1.535(3)

1.419(3)

1.149(3)

	11117(3)	. ((,) (2)	1.505(2)
C(4)-C(5)	1.529(3)	C(14)-N(5)	1.510(8)
C(4)-C(9)	1.554(3)	N(5)—C(15)	1.428(8)
C(9)-C(10)	1.536(3)	C(15)-C(16)	1.52(1)
C(10)-C(11)	1.502(5)	C(16)-O	1.48(1)
C(11)-C(12)	1.515(5)	O-C(17)	1.397(9)
C(12)-C(13)	1.531(4)	C(17)-C(18)	1.454(8)
C(13)-C(4)	1.550(4)	C(18)-N(5)	1.456(6)
C(5)-C(8)	1.407(3)		
Angle	ω/deg	Angle	ω/deg
C(16)-O-C(17)	108.4(7)	C(4)-C(5)-C(6)	122.2(2)
C(2)-N(1)-C(6)	121.8(2)	C(4)-C(5)-C(8)	121.7(2)
C(14)-N(5)-C(15)	108.2(5)	C(6)-C(5)-C(8)	116.0(2)
C(14)-N(5)-C(18)	112.9(4)	N(1)-C(6)-N(4)	115.0(2)
C(15)-N(5)-C(18)	112.3(5)	N(1)-C(6)-C(5)	120.5(2)
S-C(2)-N(1)	116.0(1)	N(4)-C(6)-C(5)	124.5(2)
S-C(2)-C(3)	125.6(1)	N(2)-C(7)-C(3)	178.8(3)
N(1)-C(2)-C(3)	118.4(2)	N(3)-C(8)-C(5)	176.1(3)
C(2)-C(3)-C(4)	124.8(2)	C(4)-C(9)-C(10)	115.2(2)
C(2)-C(3)-C(7)	116.8(2)	C(9)-C(10)-C(11)	112.9(2)
C(4)-C(3)-C(7)	117.9(2)	C(10)-C(11)-C(12)	111.5(3)
C(3)-C(4)-C(5)	106.7(2)	C(11)-C(12)-C(13)	111.0(3)
C(3)-C(4)-C(9)	109.2(2)	C(4)-C(13)-C(12)	113.0(2)
C(5)-C(4)-C(9)	112.4(2)	N(5)-C(15)-C(16)	107.2(6)
C(3)-C(4)-C(13)	108.8(2)	O-C(16)-C(15)	115.0(7)
C(5)-C(4)-C(13)	110.7(2)	O-C(17)-C(18)	111.7(5)
C(9)-C(4)-C(13)	109.0(2)	N(5)-C(18)-C(17)	112.0(4)

conjugated cyano group in the region of 2250 cm⁻¹. The ¹H NMR spectra of compounds 9 showed signals for the protons of the amino group as a singlet at δ 5.80, the signals of the N-H proton of the dihydropyridine nucleus of the amino and imino forms as singlets at δ 8.00 and 8.50, respectively, and the signal of the proton of the =NH group as a singlet at δ 9.38–9.46 (Tables 3 and 4).

Table 2. Atomic coordinates ($\times 10^4$; for H, $\times 10^3$) in the structure of 6

Atom	x	у	ζ
<u> </u>	2082(1)	3052(1)	1978(1)
0	835(6)	1975(4)	4646(3)
N(1)	3772(3)	2079(1)	3210(1)
N(2)	139(4)	1913(1)	-110(2)
N(3)	3449(6)	-248(1)	4024(2)
N(4)	4952(3)	1405(1)	4579(1)
N(5)	1225(5)	1744(2)	6714(3)
C(2)	2620(3)	2200(1)	2276(1)
C(3)	2036(3)	1654(1)	1666(1)
C(4)	2653(3)	910(1)	1912(1)
C(5)	3351(3)	869(1)	3044(2)
C(6)	4023(3)	1430(1)	3614(1)
C(7)	978(3)	1802(1)	688(2)
C(8)	3421(5)	245(1)	3559(2)
C(9)	941(4)	427(1)	1511(2)
C(10)	1441(5)	-335(1)	1474(2)
C(11)	3064(6)	-461(2)	984(2)
C(12)	4780(5)	-30(2)	1449(3)
C(13)	4278(4)	732(1)	1391(2)
C(14)	2363(8)	1615(4)	7761(6)
C(15)	1444(10)	1167(3)	6115(7)
C(16)	346(12)	1320(7)	5061(10)
C(17)	605(7)	2501(4)	5294(5)
C(18)	1730(6)	2384(2)	6298(4)
H(1)	397(4)	240(1)	359(2)
H(4a)	530(4)	106(1)	484(2)
H(4b)	519(4)	174(1)	486(2)
H(5)	-4(4)	177(1)	678(2)
H(9a)	32(4)	59(1)	81(2)
H(9b)	-2(4)	53(1)	194(2)
H(10a)	28(4)	-59(1)	114(2)
H(10b)	178(4)	-52(1)	219(2)
H(lla)	335(4)	-96(2)	100(2)
H(11b)	257(4)	-34(1)	32(2)
H(12a)	582(4)	-9(1)	110(2)
H(12b)	529(4)	-16(1)	219(2)
H(13a)	541(4)	101(1)	170(2)
H(13b)	386(4)	87(1)	72(2)
H(14a)	199(4)	110(2)	791(2)
H(14b)	382(4)	167(1)	772(2)
H(14c)	231(5)	206(2)	783(2)
H(15a)	93(4)	71(2)	641(2)
H(15b)	290(4)	110(1)	607(2)
H(16a)	57(5)	112(2)	476(3)
H(16b)	-61(5)	135(2)	502(3)
H(17a)	-70(4)	254(1)	538(2)
H(17b)	106(4)	293(2)	501(2)
H(18a)	307(4)	233(1)	628(2)
H(18b)	145(4)	277(1)	696(2)

Scheme 1

8, 9: Hal = Br, R =
$$4-BrC_6H_4CO$$
 (a), Et (d), $4-ClC_6H_4CO$ (e), $2-MeC_6H_4$ (h), $4-MeC_6H_4CO$ (i); Hal = Cl, R = Ph (b), $4-BrC_6H_4NHCO$ (f), PhNHCO (g); Hal = I, R = Me (c), H (k).

Alkylation of thione 7 with 1,2-dibromoethane yielded 5-amino-6,8-dicyano-spiro-2,3-dihydrothiazolo-[3,2-a]pyridine-7,1'-cyclohexane (10), whereas treatment of substituted 2-(4-bromobenzoylmethylthio)dihydropyridine (9a) with a 10% aqueous KOH solution yielded thienopyridine (11). The structures of compounds 10 and 11 are consistent with the data of spectral and physicochemical studies (see Tables 3 and 4).

The dihydropyridine heterocycle (see Fig. 1 and Table 1) has a boat conformation; the N(1) and C(4) atoms deviate from the plane through the other atoms of

the "bottom" (planar to within ± 0.022 Å) by 0.116 and 0.296 Å, respectively, which corresponds to the fold of the ring along the C(2)...C(6), C(3)...C(5), and N(1)...C(4) lines by 10.0°, 18.9°, and 18.8°, respectively. Previously, we have established an analogous conformation of the 1,4-dihydropyridine ring in the molecule of morpholinium 5-acetyl-3-cyano-6-methyl-4-(2-nitrophenyl)-1,4-dihydropyridine-2-thiolate (12).

As can be seen from Fig. 1, the cyclohexane spiro substituent has a chair conformation. The C(4) and C(11) atoms deviate from the plane through the C(9),

Table 3. Yields, melting points, and the data of IR spectroscopy for spirodihydropyridines **9a**—k

Com-	Yield (%)	M.p./°C	. v/cm ⁻¹				
pound	(method A/B)	(solvent for crystallization)	NH, NH ₂	CN	СО		
9a	69/65	198-200	3195, 3305,	2185	1710		
		(EtOH)	3364				
9b	92/84	158-160	3390, 3474	2172,			
	,	(EtOH)		2190			
9c	80/75	141-143	3175-3390	2178			
	•	(EtOH)					
9d	77/85	147—149	3183, 3190,	2190,			
	,	(EtOH)	3308, 3390	2248			
9e	81/69	179—181	3180, 3302,	2188,	1684		
		(AcOH)	3384	2250			
9f	74/70	199-201	3182, 3304,	2189,	1690		
		(EtOH)	3390, 3496	2245			
9h	63/61	158-160	3150, 3328,	2188,			
		(Bu ⁿ OH)	3445	2196			
9g	71/84	203-205	3155, 3422	2170,	1682		
_	•	(EtOH)		2253			
9i	63/60	189-191	3335-3450	2190,	1674		
	•	(EtOH)		2215			
9k	90/88	143-145	3304-3400	2180,			
	•	(EtOH)		2242			

Table 4. Data of ¹H NMR spectroscopy and elemental analysis of spirodihydropyridines 9a-k

Com-	i	δ						Found (%) Calculated			Molecular formula
	(SH ₂) ₅ (m)	NH_2 , = NH (s)	N—H (s)	C(3)H (s)	SCH ₂	R	С	Н	Ν	S	
9a	1.31—	5.82,	8.17,	4.19	4.66 (s)	7.85 (m)	54.20	4.41	12.58	7.34	C ₂₀ H ₁₉ BrN ₄ OS
	1.77	9.49	8.51				54.18	4.32	12.64	7.23	
9b	1.15—	5.78,	8.17,		4.23 (s)	7.29 (m)	<u>67.78</u>	<u>5.93</u>	<u> 16.71</u>	9.47	$C_{19}H_{20}N_4S$
	1.85	9.46	8.71				67.83	5.99	16.65	9.53	
9c	1.20—	5.82,	8.07,	4.19	2.99 (q)	1.19 (t)	61.33	6.57	<u> 20.47</u>	11.65	$C_{14}H_{18}N_4S$
	1.76	9.38	8.58				61.28	6.61	20.42	11.69	
9d	1.15—	5.80,	8.06,	4.17	2.93 (t)	0.92 (m)	62.52	<u>7.03</u>	19.38	11.18	$C_{15}H_{20}N_4S$
	1.80	9.38	8.55				62.47	6.99	19.43	11.12	
9e	1.3.	5.80,	s.15,	17	4.66 (s)	7.60 and	60.19	4.84	13.89	8.08	C ₂₀ H ₁₉ CIN ₄ OS
	1.76	9.47	8.48			7.99 (d)	60.22	4.80	14.05	8.04	
9f	1.33—	5.84,	8.22,	4.22	3.91 (s)	7.51 (d),	52.37	<u>4.45</u>	15.35	6.87	$C_{20}H_{20}BrN_5OS$
	1.76	9.65	8.61			10.16 and	52.41	4.40	15.28	7.00	
						10.40 (s)					
9h	1.00-	5.84,	8.15,		4.22 (s)	2.28 (s),	68.50	6.29	16.03	9.18	$C_{20}H_{22}N_4S$
	1.85	9.50	8.76			7.12 (m)	68.54	6.33	15.98	9.15	20 22 .
9g	1.34-	5.89,	8.25,	4.25	3.93 (s)	7.05—	63.42	5.64	18.31	8.39	$C_{20}H_{21}N_5OS$
Ü	1.78	9.74	8.66			7.57 (m),	63.30	5.58	18.45	8.45	20 21 3
						10.05 and					
						10.31 (s)					
9i	1.32—	5.81,	8.14,	4.20	4.67 (s)	2.39 (s),	66.68	5.79	14.85	8.42	$C_{21}H_{22}N_4OS$
	1.76	9.46	8.54	= -		7.35 and	66.64	5.86	14.80	8.47	21 22 7
	, 0					7.87 (d)					
9k	1.31-	5.06,	8.11,	4.17	2.37 and	(=/	<u>59.93</u>	6.21	21.46	12.37	$C_{13}H_{16}N_4S$
· ••	1.80	9.28	8.62		2.46 (s)		59.97	6.19	21.52	12.32	-13104-

C(10), C(12), and C(13) atoms (to within ± 0.032 Å) by 0.625 and -0.646 Å, respectively. This ring has an analogous conformation in the molecules of 4-amino-5-

cyano-6-phenyl-2-cyclohexanespiro-1,3-dithia-4-cyclohexene 10 and 2,5-dicyclohexanespiro-1,3-dithiacyclohexane. 11 It should be noted that the forced shortened

nonbonded intramolecular contacts (C(7)...C(9), 2.918(3) Å; C(7)...C(13), 3.153(4) Å; C(8)...C(9), 2.999(3) Å; C(8)...C(10), 3.116(3) Å; C(8)...C(12), 3.321(4) Å; and C(8)...C(13), 3.327(4) Å (the double van der Waals radius of the C atom is 3.40 Å¹²)) affect the conformation of the cyclohexane substituent (the dihedral angle between the planar fragments of the rings in the anion is 105.5°).

As in 12, the negative charge in the anion of salt 6 is formally located on the sulfur atom as evidenced by the C(2)—S bond length (1.737(2) Å; 1.725(3) Å in 12), which is substantially larger than that of the conjugated multiple C=S bond (1.660(4) Å and 1.666(5) Å) in substituted pyridine-2(1 H)-thiones, ^{13,14} but is smaller than the C—S bond length (1.753(4) Å) in 2-allylthio-3-cyano-5-ethoxycarbonyl-6-methyl-4-phenyl=1,4-dihydropyridine. ^{15,16}

As noted for the anion of salt 12, conjugation in the planar $S-C(2)=C(3)-C(7)\equiv N(2)$ and $N(4)-C(6)=C(5)-C(8)\equiv N(3)$ fragments in the anion of salt 6 also results in a substantial redistribution of bond lengths (shortening of single bonds and lengthening of double bonds) compared to the standard values.¹⁷

The heterocycle of the N-methylmorpholinium cation adopts a chair conformation. The O and N(5) atoms deviate from the plane through the C(15), C(16), C(17), and C(18) atoms (planar to within ± 0.014 Å) by 0.637 and -0.622 Å, respectively.

In the crystal of 6, cations and anions of the salt are linked in layers parallel to the bc plane via intermolecular hydrogen bonds: N(1)-H(1)...N(2) (0.5 + x, 0.5 - y, 0.5 + z) [N(1)...N(2) 3.029(3), N(1)-H(1) 0.81(3), H(1)...N(2) 2.24(2) Å, the N(1)-H(1)...N(2) angle is $164(2)^{\circ}$], N(4)-H(4a)...N(3) (1 - x, 1 - y, 1 - z)

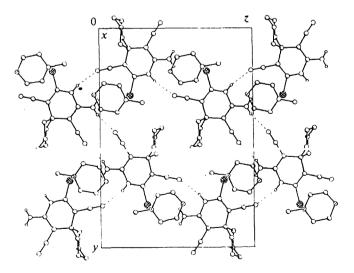


Fig. 2. Projection of the crystal structure of 6 onto the ab plane. Intermolecular N—H...N hydrogen bonds are indicated by dashed lines.

[N(4)...N(3) 3.023(3), N(4)—H(4a) 0.77(3), H(4a)...N(3) 2.28(2) Å, the N(4)—H(4a)...N(3) angle is $164(2)^{\circ}$], and N(5)—H(5)...S (-0.5 + x, 0.5 - y, 0.5 + z) [N(5)...S 3.124(3), N(5)—H(5) 0.94(3), H(5)...S 2.19(2) Å, the N(5)—H(5)...S angle is $173(2)^{\circ}$] (Fig. 2). The second hydrogen atom of the N(4)H₂ group is involved in a weak intermolecular contact H(4b)...N(2) (0.5 + x, 0.5 - y, 0.5 - z) with the distance of 2.63(3) Å, which is comparable to the sum of the van der Waals radii (2.75 Å).¹²

Experimental

The IR spectra were recorded on an IKS-29 spectrophotometer as Nujol mulls. The ¹H NMR spectra were measured on a Bruker WP 100 SY instrument (100 MHz) in DMSO-d₆ relative to TMS. The purity of the compounds was checked by TLC on Silufol UV-254 plates in a 3:5 acetone—hexane system.

N-Methylmorpholinium 6-amino-3,5-dicyano-1,4-dihydropyridine-4-spiro(cyclohexane)-2-thiolate (6). Method A. N-Methylmorpholine (2 mL, 20 mmol) was added with stirring to a mixture of 1 (10 mmol) and 2 (10 mmol) in absolute ethanol (15 mL) at 20 °C. After 48 h, the colorless crystals that formed were filtered off and washed with absolute ethanol and hexane. The yield was 2.5 g (71%), m.p. 133−135 °C. Found (%): C, 58.81; H, 7.21; N, 20.20; S, 9.19. $C_{17}H_{25}N_5OS$. Calculated (%): C, 58.76; H, 7.25; N, 20.16; S, 9.23. IR, v/cm⁻¹: 3310, 3405 (NH, NH₂), 1600 (δ (NH₂)), 2190 (C≡N). ¹H NMR, δ : 8.05 (s, 1 H, NH); 5.43 (s, 2 H, NH₂); 3.74 (m, 4 H, CH₂NCH₂); 3.02 (m, 4 H, CH₂OCH₂); 2.68 (s, 3 H, CH₃); 1.54 (m, 10 H, (CH₂)₅).

Method B. N-Methylmorpholine (2 mL, 20 mmol) was added with stirring to a suspension of 1 (10 mmol) and 3 (10 mmol) in absolute ethanol (15 mL) at 20 °C. After 24 h, the precipitate that formed was separated and washed with ethanol and hexane. Compound 6 was obtained in a yield of 2.7 g (78%). The melting point and the ¹H NMR spectrum of this compound were identical to those of the compound obtained by method A.

Crystals of compound 6 are monoclinic, at 20 °C: a =7.230(2) Å, b = 19.537(3) Å, c = 13.788(4) Å, $\beta = 103.25(2)^{\circ}$, V= 1896(2) Å³, d_{calc} = 1.224 g cm⁻³, Z = 4, $C_{17}H_{25}N_5OS$, M = 347.41, space group $P2_1/n$. The unit cell parameters and intensities of 4265 independent reflections were measured on a four-circle automated Siemens P3/PC diffractometer (λ(Mo-Kα) radiation, graphite monochromator, $\theta/2\theta$ scanning technique to $\theta_{\text{max}} = 28^{\circ}$). The structure was solved by the direct method. All nonhydrogen atoms were revealed and refined anisotropically by the full-matrix least-squares method using 3847 reflections with $I > 3\sigma$ (I). All hydrogen atoms were located from difference Fourier syntheses and refined isotropically. The final values of the R factors were as follows: R = 0.057 and $R_w = 0.057$. All calculations were carried out using the SHELXTL PLUS program¹⁸ (the PC version). Atomic coordinates are given in Table 2 (thermal parameters of the atoms may be obtained from the authors).

6-Amino-3,5-dicyano-1,4-dihydropyridine-4-spirocyclo-hexane-2-(1H)-thiol (7). A 10% HCl solution was added dropwise with stirring to a suspension of 6 (10 mmol) in ethanol (15 mL) until pH became 5. Then the reaction mixture was filtered and kept at 20 °C for 24 h. The yellow powder that formed was separated and washed with ethanol and hexane.

The yield was 2.3 g (92%), m.p. 194–196 °C (from ethanol) (cf. Ref. 8: m.p 168–169 °C). Found (%): C, 58.60; H, 5.60; N, 22.81; S, 12.99. $C_{12}H_{14}N_4S$. Calculated (%): C, 58.51; H, 5.73; N, 22.74; S, 13.02. IR, v/cm^{-1} : 3210, 3334, 3448 (NH, NH₂); 1646 (δ (NH₂)); 2195 and 2256 ($C\equiv$ N). ¹H NMR, δ : 12.13 (s, 1 H, NH); 6.27 (s, 2 H, NH₂); 4.60 (m, 0.5 H, C(3)); 4.06 (m, 0.5 H, SH); 1.20–1.75 (m, 10 H, (CH₂)₅).

6-Amino-3,5-dicyano-2-R-methylthio-1,4-dihydropyridine-4-spirocyclohexanes (9a—k). Method A. Halide 8 (10 mmol) was added with stirring to a suspension of salt 6 (10 mmol) in DMF (10 mL). Then the mixture was stirred for 4 h and diluted with water (10 mL). The precipitate was separated and washed with water, ethanol, and hexane. Compounds 9 were obtained (see Tables 3 and 4).

Method B. A 10% aqueous KOH solution (5.6 mL, 10 mmol) was added with stirring to a solution of thione 7 (10 mmol) in DMF. After 5 min, halide 8 was added. The reaction mixture was stirred for 4 h and diluted with water (10 mL). The precipitate was washed sequentially with water, ethanol, and hexane (see Tables 3 and 4).

5-Amino-6,8-dicyano-spiro-(2,3-dihydrothiazolo[3,2-a]pyridine-7,1'-cyclohexane) (10). A 10% aqueous KOH solution (5.6 mL, 10 mmol) was added with stirring to a solution of thione 7 (10 mmol) in DMF (8 mL), and then 1,2-dibromoethane (0.86 mL, 10 mmol) was added. After 30 min, more aqueous 10% KOH solution (5.6 mL, 10 mmol) was added to the reaction mixture. The reaction mixture was stirred for 8 h and then diluted with water (15 mL). The precipitate was separated and washed sequentially with water, ethanol, and hexane. The yield was 1.85 g (68%), m.p. 156-158 °C (from BuⁿOH). Found (%): C, 61.60; H, 6.11; N, 20.64; S, 11.65. C₁₄H₁₆N₄S. Calculated (%): C, 61.74; H, 5.92; N, 20.57; S, 11.77. IR, v/cm⁻¹: 3345, 3420 (NH₂); 1635 $(\delta(NH_2))$; 2175 (CN). ¹H NMR, δ : 6.24 (s, 2 H, NH₂); 4.09 (t, 2 H, SCH₂); 3.38 (t, 2 H, NCH₂); 1.40-1.64 (m, 10 H, $(CH_2)_5$).

3,6-Diamino-2-(4-bromobenzoyl)-5-cyano(spirothieno[2,3-b]-4,7-dihydropyridine-4-spirocyclohexane) (11). A 10% aqueous KOH solution (5.6 mL, 10 mmol) was added with stirring to a solution of compound 9a (10 mmol) in DMF (10 mL). After 4 h, the reaction mixture was diluted with water (10 mL). The precipitate was filtered off and washed with water, ethanol, and hexane. The yield was 2.4 g (54%), m.p. 165-167 °C (from AcOH). Found (%): C, 54.21; 11, 4.38; N, 12.63; S, 7.28. $C_{20}H_{19}BrN_4OS$. Calculated (%): C, 54.18; H, 4.32; N, 12.64; S, 7.23. IR, v/cm^{-1} : 3330, 3365, 3428 (NH, NH₂); 1675 (C=O); 2195 and 2254 (C=N). ¹H NMR, δ : 8.30 (s, 1 H, NH); 7.81 (s, 0.5 H, N-H); 7.75 (s, 3 H, NH₂); 7.45 (d, 2 H, Ar); 7.61 (d, 2 H, Ar); 4.44 (s, 0.5 H, C(5)H); 1.49 (m, 10 H, (CH₂)₅).

References

- A. A. Krauze, R. O. Vitolinya, G. V. Zarin'sh, Yu. E. Pelcher, Z. A. Kalme, A. A. Kamenis, and G. Ya. Dubur, *Khim.-farm. Zh.*, 1985, 19, 540 [*Pharm. Chem. J.*, 1985, 19 (Engl. Transl.)].
- A. A. Krauze, R. O. Vitolinya, M. R. Romanova, and G. Ya. Dubur, Khim.-farm. Zh., 1988, 22, 548; 955 [Pharm. Chem. J., 1988, 22 (Engl. Transl.)].
- 3. A. P. Phillip, J. Am. Chem. Soc., 1949, 71, 4003.
- 4. B. Loev, S. Ehrreich, and R. E. Tedeschi, *J. Pharm. Pharmacol.*, 1979, **24**, 917.
- F. Bossert, H. Meyer, and G. Wehinger, Angew. Chem., 1981, 93, 755.
- V. V. Kuznetsov, Khim.-farm. Zh., 1991, 25, 61 [Pharm. Chem. J., 1991, 25 (Engl. Transl.)].
- 7. V. P. Litvinov, *Phosphorus*, *Sulfur*, and *Silicon*, 1993, 74, 139.
- 8. F. F. Abdel-Latif, Pharmazie, 1990, 45, 283.
- V. N. Nesterov, V. E. Shklover, Yu. T. Struchkov, Yu. A. Sharanin, A. M. Shestopalov, and L. A. Rodinovskaya, Acta Crystallogr., 1985, C41, 1191.
- Yu. A. Sharanin, V. P. Litvinov, A. M. Shestopalov, V. N. Nesterov, Yu. T. Struchkov, V. E. Shklover, V. K. Promonenkov, and V. Yu. Mortikov, Izv. Akad. Nauk SSSR, Ser. Khim., 1985, 1768 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1985, 34, 1619 (Engl. Transl.)].
- K. Stadnicka and T. Lebioda, Acta Crystallogr., 1982, B38, 985.
- 12. A. Bondi, J. Phys. Chem., 1966, 70, 3006.
- Yu. A. Sharanin, A. M. Shestopalov, L. A. Rodinovskaya,
 V. N. Nesterov, V. E. Shklover, Yu. T. Struchkov, V. K.
 Promonenkov, and V. P. Litvinov, Zh. Org. Khim., 1986,
 22, 2600 [J. Org. Chem. USSR, 1986, 22 (Engl. Transl.)].
- Yu. A. Sharanin, V. K. Promonenkov, A. M. Shestopalov, L. A. Rodinovskaya, V. N. Nesterov, S. N. Melenchuk, V. E. Shklover, and Yu. T. Struchkov, Zh. Org. Khim., 1989, 25, 622 [J. Org. Chem. USSR, 1989, 25 (Engl. Transl.)].
- 15. T. P. E. Auf der Heyde, H.-B. Burgi, and V. E. Shklover, Acta Crystallogr., 1991, C47, 566.
- 16. V. N. Nesterov, V. E. Shklover, Yu. T. Struchkov, Yu. A. Sharanin, M. P. Goncharenko, and V. D. Dyachenko, Izv. Akad. Nauk SSSR, Ser. Khim., 1991, 521 [Bull. Acad. Sci. USSR, Div. Chem. Sci., 1991, 40, 453 (Engl. Transl.)].
- F. H. Allen, O. Kennard, D. G. Watson, L. Brammer,
 A. G. Orpen, and R. Taylor, J. Chem. Soc., Perkin Trans.
 1987
- W. Robinson and G. M. Sheldrick, in *Crystallographic Computing Techniques and New Technologies*, Oxford Univ. Press, Oxford, 1988, 366.